Research Article

Comprehension of Postmetallization Annealed MOCVD-TiO₂ **on** (NH₄)₂**S Treated III-V Semiconductors**

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The electrical characteristics of TiO₂ films grown on III-V semiconductors (e.g., p-type InP and GaAs) by metal-organic chemical vapor deposition were studied. With $(NH_4)_2S$ treatment, the electrical characteristics of MOS capacitors are improved due to the reduction of native oxides. The electrical characteristics can be further improved by the postmetallization annealing, which causes hydrogen atomic ion to passivate defects and the grain boundary of polycrystalline TiO₂ films. For postmetallization annealed TiO₂ on $(NH_4)_2S$ treated InP MOS, the leakage current densities can reach 2.7×10^{-7} and 2.3×10^{-7} A/cm² at ±1 MV/cm, respectively. The dielectric constant and effective oxide charges are 46 and 1.96×10^{12} C/cm², respectively. The interface state density is 7.13 × 10^{11} cm⁻² eV⁻¹ at the energy of 0.67 eV from the edge of valence band. For postmetallization annealed TiO₂ on $(NH_4)_2S$ treated GaAs MOS, The leakage current densities can reach 9.7×10^{-8} and 1.4×10^{-7} at ±1 MV/cm, respectively. The dielectric constant and effective oxide charges are 66 and 1.86×10^{12} C/cm², respectively. The interface state density is 5.96×10^{11} cm⁻² eV⁻¹ at the energy of 0.7 eV from the edge of valence band.

1. Introduction

Due to its high electron mobility and direct energy band gap compared with Si, much attention has been focused on III-V compound semiconductor (e.g., InP and GaAs) high-speed devices. Usually, the metal-semiconductor fieldeffect transistor (MESFET) is one of III-V main highspeed devices due to the lack of high quality of oxide on it. The main disadvantage of MESFET is the high gate currents of Schottky contact under the positive bias of several tenths of a volt, which severely limits the maximum drain currents, the lower noise margin, and the less flexibility of the circuit design. Metal-oxide-semiconductor field-effect transistor (MOSFET) can alleviate these problems. Many high-k dielectrics, such as Al_2O_3 [1], TiO_2 [2], Ga_2O_3 [3], HfAlO [4], and HfO₂ [5], are currently being explored on III-V substrates. For high-k dielectrics, the same gate capacitance per unit area can be realized using a much thicker gate materials and results in less tunneling leakage current. Of various high-k materials, TiO_2 has generated much interest

as it offers a large dielectric constant (*k* value 4–86) [6] and a higher transconductance of MOSFET is expected [7].

High dielectric constant polycrystalline TiO_2 films were prepared by metal-organic chemical vapor deposition (MOCVD) [8], sol-gel [9], and sputtering [10]. MOCVD-TiO₂ was used in this study because of its simple process and higher quality. Usually, the leakage current of MOCVD-TiO₂ on III-V is high from the high interface state (D_{it}) [11, 12] and the polycrystalline grain boundary [13, 14]. From previous studies [13, 14], the high D_{it} from native oxides on III-V surface can be removed by (NH₄)₂S treatment. It can also passivate the surface dangling bonds of III-V surface and prevent it from oxidizing.

The low temperature postmetallization annealing (PMA) is an effective process to reduce the oxide charge density and the D_{it} in SiO₂/Si metal-oxide-semiconductor (MOS) technology [15, 16]. The mechanism of PMA process is from the reaction between the aluminum contact and hydroxyl groups existed on SiO₂ films surface resulting in hydrogen atomic ions diffusing through the oxide and passivate the



FIGURE 1: (a) Images of SEM and HRTEM of TiO₂/S-InP and (b) SEM cross section of TiO₂/S-GaAs substrate.



FIGURE 2: SIMS depth profiles for (a) TiO₂/S-InP and (b) PMA ($350^{\circ}C$)-TiO₂/S-InP.



FIGURE 3: SIMS depth profiles for (a) TiO₂/S-GaAs and (b) PMA (350°C)-TiO₂/S-GaAs.

In-O

443

444

Binding energy (eV)

(a)

445

446

442

Intensity (a.u.)

440

441

MOS structures	Dielectric constant	Leakage current at 1	MV/cm	Interface state density	$\Delta V_{\rm FB}$ of hysteresis loop
PMA(350°C)-TiO ₂ /S-InP	44	2.7×10^{-7} and 2.3×10^{-7}	0^{-7}A/cm^2	$7.13 imes 10^{11} \mathrm{cm}^{-2} \mathrm{eV}^{-1}$	17 mV
PMA(350°C)-TiO ₂ /S-GaAs	66	9.7×10^{-8} and 1.4×10^{-7}		$5.96 imes 10^{11} \mathrm{cm}^{-2} \mathrm{eV}^{-1}$	9 mV
In 3d _{5/2}	/ In-P		In 3d _{5/2}	∩ ^{In-P}	

Intensity (a.u.)

440

441

442

In-O

443

444

Binding energy (eV) (b)

445

TABLE 1: Electrical characteristics by PMA(350°C)-TiO₂/InP and PMA(350°C)-TiO₂/GaAs.



448

447







In-S

446

447

448

FIGURE 6: Leakage current densities of TiO_2/InP with and without $(NH_4)_2S$ treatments and PMA- TiO_2/S -InP at different PMA temperatures.

oxide traps [15–17]. From our previous study [18], the PMA was used to reduce the leakage current from the defects and grain boundary of polycrystalline TiO_2 films grown on silicon. Both treatments also show the same function on high-*k*/III-V. In this study, we try to review the improvement of electrical characteristics of TiO_2/InP and $TiO_2/GaAs$ by the combination of $(NH_4)_2S$ and PMA treatments (PMA- TiO_2/S -InP (GaAs)).

2. Experimental

Zn doped p-type (100) InP and GaAs with carrier concentration of 5×10^{16} and 6×10^{16} cm⁻³ were used as the substrates. These substrates were degreased in solvent and followed by chemical etching in a solution (H₂SO₄:H₂O₂:H₂O = 5:1:1) for 3 min and then rinsed in deionized water.



FIGURE 7: Leakage current densities of $TiO_2/GaAs$ with and without $(NH_4)_2S$ treatments and without $(NH_4)_2S$ treatment and PMA- TiO_2/S -GaAs at different PMA tempreatures.



FIGURE 8: C-V characteristics of TiO₂/InP with and without (NH₄)₂S treatments and PMA-TiO₂/S-InP at different PMA temperatures.

After cleaning, substrates were immediately dipped into $(NH_4)_2S$ solution for 40 min at 50°C and then blow-dried with nitrogen gas. After $(NH_4)_2S$ treatment, substrates were thermally treated at 220°C in a nitrogen atmosphere for 10 min in order to desorb the excess of weakly bonded sulfur and were ready for MOCVD-TiO₂ growth.

Polycrystalline TiO₂ thin films were grown on substrates by a horizontal cold-wall MOCVD system. Tetraisopropoxytitanium (Ti(i-OC₃H₇)₄) was used as a Ti precursor and kept at 24°C. Nitrogen was used as the carrier gas and its flow rate was 10 sccm. Nitrous oxide gas (N₂O) was used as an oxidizing agent and its flow rate was 100 sccm. Molybdenum was used as the oxidation-resist susceptor. The reactor pressure was kept at 5 Torr during the growth. The growth temperature was kept at 400°C for 5 min. The chemical reaction steps during the deposition of TiO₂ on substrate in MOCVD system are as follows.

In PMA procedure, aluminum (Al) was deposited upon the TiO₂ films as the cap layer. Then, these films were annealed in nitrogen ambient for 10 min at the temperature of 300, 350, and 400°C, respectively. Finally, the Al was etched away with an etching solution (H₃PO₄:HNO₃:CH₃COOH:H₂O = 73:4:3.5:19.5).

A metal-oxide-semiconductor (MOS) structure was used to examine the electrical characteristics. In-Zn alloy (In 90% and Zn 10%) was evaporated on the III-V back side for ohmic contact and then thermally annealed at 400°C for 3 min in nitrogen atmosphere. The ohmic contact was confirmed by the current-voltage characteristics. Then, Al was evaporated on TiO₂ films as the top contact with the area of 7.07 × 10^{-4} cm². Scanning electron microscopy (SEM) was used to examine the thickness of TiO₂ film. A HP4145B semiconductor-parameter analyzer was used for currentvoltage (I-V) characterization. A high frequency (1 MHz)HP4280A capacitance-voltage (C-V) meter was used for C-V characterization scanned from accumulation region to inversion region. The DC bias was swept by 1/30 V/sec. The $D_{\rm it}$ was derived from C-V curves by Terman method [19], which can provide a good evaluation [20] of the D_{it} higher than $10^{10} \text{ cm}^{-2} \text{ eV}^{-1}$ with 10% error [21, 22].

3. Results and Discussion

The SEM cross section of TiO_2/S -InP is shown in Figure 1(a) and the thickness of TiO_2 film is 59 nm. The SEM picture shows that there is an interfacial layer. From the image of high-resolution transmission electron microscopy (HRTEM) shown in the inset in Figure 1(a), the interfacial layer is 2.5 nm. It is from the interdiffusion between TiO_2 and InP examined by SIMS depth profile shown in Figure 2(a). The SEM cross section of TiO_2/S -GaAs is shown in Figure 1(b) and the thickness of TiO_2 film is 62 nm. The SEM picture also shows an interfacial layer. The mechanism is the same as TiO_2/S -InP and examined by the SIMS depth profile shown in Figure 3(a). It indicates that a low temperature growth process is essential for decreasing the interfacial layer for ultrathin TiO_2 film. Atomic layer deposition (ALD) may be the candidate.

The X-ray photoelectron spectroscopy (XPS) core level spectra of In $3d_{5/2}$ of InP without and with $(NH_4)_2S$ treatments are shown in Figures 4(a) and 4(b), respectively. For InP without $(NH_4)_2S$ treatment, the strong In $3d_{5/2}$ XPS peak at 444.7 eV can be attributed to In-P bond [23] and the peak at 443.7 eV is from In-O bond [24]. For InP with $(NH_4)_2S$ treatment, a new strong peak at 444.9 eV [25] and a small satellite peak at 443.9 eV are from In-S and In-O, respectively. The strong In-S bond shows that In empty dangling bond is passivated by S. The much weaker



FIGURE 9: (a) Dielectric constant and (b) effective oxide charges of TiO₂/S-InP and PMA-TiO₂/S-InP at different PMA temperatures.

In-O peak indicates that native oxides are significantly reduced after $(NH_4)_2S$ treatment. It suggests the $(NH_4)_2S$ treatment can not completely remove native oxides on III-V semiconductors. It needs a further investigation.

In order to double check the function of $(NH_4)_2S$ passivation, the photoluminescence (PL) spectra of GaAs and S-GaAs are shown in the inset of Figure 5. The lower PL intensity for GaAs without $(NH_4)_2S$ treatment shows a high surface recombination velocity [26, 27]. The PL intensity is much improved for GaAs with $(NH_4)_2S$ treatment. It indicates that the $(NH_4)_2S$ passivation is an effective way to reduce the surface recombination velocity [28]. It supports that $(NH_4)_2S$ treatment can remove native oxides and prevent III-V surface from oxidizing.

The leakage current densities of TiO₂ film deposited on InP substrate with and without (NH₄)₂S treatments are shown in Figure 6. It also shows the leakage current densities of PMA-TiO₂/S-InP treated at the PMA temperatures of 300, 350, and 400°C. The leakage current densities of TiO₂/InP are 0.1 and 0.57 A/cm² at ± 1 MV/cm as shown in Figure 6(a). The high leakage currents are mainly from the high density of defects in the grain boundary of polycrystalline TiO₂ film [11, 12] and the high interface states at TiO_2/InP interface due to InP native oxide [13, 14]. For TiO₂/S-InP as shown in Figure 6(b), the leakage current densities are 4.56×10^{-6} and 0.16 A/cm^2 at $\pm 1 \text{ MV/cm}$, respectively. The leakage currents are mainly from D_{it} and grain boundary of TiO₂ film. After (NH₄)₂S treatment, higher quality TiO₂ film can be deposited on reconstructed InP surface [14]. The leakage current is much improved under positive bias from the reduction of D_{it} . However, only one order improvement of the leakage current under negative bias is from grain boundary.

The leakage current density can be further improved by PMA treatment as shown in Figures 6(c)-6(e). The lowest leakage current densities of PMA-TiO₂/S-InP can reach 2.7 \times 10⁻⁷ and 2.3 \times 10⁻⁷ A/cm² at ±1 MV/cm at the PMA treatment of 350°C as shown in Figure 6(d). After PMA process, the thickness of the interfacial layer does not change and is examined from SIMS depth profiles as shown in Figures 2(a) and 2(b). Therefore, the improvement of leakage current is not from the increase of interfacial layer thickness after PMA. Figure 2(b) shows that H atoms are uniformly distributed in the whole TiO₂ film due to H fast diffusion after PMA treatment. It would diffuse along and passivate the grain boundary. At the PMA treatment of 300°C, the leakage current densities are 6.58×10^{-7} and 1.2×10^{-6} A/cm² at ± 1 MV/cm as shown in Figure 6(c). The slight increase of leakage current compared with Figure 6(d)could be from the lower PMA temperature, which cannot provide sufficient energy for H atoms for passivation. For the PMA temperature at 400° C as shown in Figure 6(e), the leakage currents are higher than that of 300 and 350°C. It is that the higher PMA temperature would destroy the H passivation and is examined by C-V characteristics.

For TiO₂ on GaAs substrate with and without (NH₄)₂S treatments, the leakage current densities are shown in Figure 7. It also exhibits the leakage current densities of PMA-TiO₂/S-GaAs treated at the PMA temperatures of 300, 350, and 400°C. The leakage current densities of TiO₂/GaAs without (NH₄)₂S treatment are 5 $\,\times\,10^{-2}$ and 5.9×10^{-1} A/cm² at ± 1 MV/cm as shown in Figure 7(a). For TiO_2/S -GaAs as shown in Figure 7(b), the leakage current densities are 2.1 $\times 10^{-6}$ and 6 $\times 10^{-3}$ A/cm² at ±1 MV/cm, respectively. The leakage current density can be further improved by PMA treatment as shown in Figures 7(c)-7(e). The leakage current densities are 4.5 $\,\times$ 10^{-7} and 4.9 $\,\times$ 10^{-7} A/cm² at ±1 MV/cm at the PMA treatment of 300°C as shown in Figure 7(c). The lowest leakage current densities of PMA-TiO₂/S-GaAs can reach 9.7 \times 10⁻⁸ and 1.4 \times 10^{-7} A/cm² at ± 1 MV/cm at the PMA treatment of 350°C as shown in Figure 7(d). For the PMA temperature at 400°C,



FIGURE 10: C-V hysteresis loops of (a) TiO₂/S-InP, (b) PMA at 300°C, (c) PMA at 350°C, and (d) PMA at 400°C.

the leakage currents are 9.5×10^{-7} and 6×10^{-6} A/cm² at ± 1 MV/cm as shown in Figure 3(e), which are higher than that of 300 and 350°C. The mechanisms are similar to InP as mentioned in the previous paragraph.

The *C*-*V* characteristics of TiO_2/InP , TiO_2/S -InP, and PMA- TiO_2/S -InP are shown in Figure 8. The *C*-*V* characteristics of TiO_2/InP show a flat curve as in Figure 8(a). It is from the high density of interface states due to the existence of native oxides on InP surface, which causes the pinning

of the surface Fermi level near the middle of the band gap [29]. Figure 8(b) shows the *C*-*V* characteristics of TiO₂/S-InP. The capacitance in the accumulation region is high due to the improved interface quality. The capacitance decay at higher negative bias is due to the high leakage current, which comes from the defects and the grain boundary of polycrystalline TiO₂ film. Sharp *C*-*V* curves PMA-TiO₂/S-InP can be obtained after PMA treatments at 300, 350, and 400°C as shown in Figures 8(c)–8(e), respectively. The ideal

FIGURE 11: Oxide trap density and mobile ion density of TiO₂/S-InP MOS structures as a function of PMA temperature.

FIGURE 12: C-V characteristics of TiO₂/GaAs with and without (NH₄)₂S treatments and PMA-TiO₂/S-GaAs at different PMA temperatures.

C-V curve is also shown in the figure as a reference. It is derived from the neglect of the effective oxide charges and the interface states, but the work function difference ($\Phi_{ms} = -1.51$ V) of metal (Al) and semiconductor (InP) is taken into account. The optimized PMA temperature is 350°C as shown in Figure 8(d), in which the stretch-out phenomenon and the flat-band voltage shift are minimized. The dielectric constants and effective oxide charges of PMA-TiO₂/S-InP films as functions of PMA temperature are shown in Figures 9(a) and 9(b), respectively. The dielectric constant increases with the PMA temperature due to the improvements of interface and film qualities. But the value decreases at PMA temperature high than 350°C. The higher PMA temperature will break the H bonds and lose the passivation function

[30, 31] and results in the increase of leakage current as shown in Figure 6. The dielectric constant and the effective oxide charges can reach 46 and 1.96×10^{12} C/cm² at the PMA temperature of 350°C. The thickness of TiO₂ film is 59 nm. The interfacial layer is very thin, which has minor effect during the extraction of dielectric constants.

Moreover, the C-V hysteresis loops as a function of PMA temperature are shown in Figure 10. The C-V hysteresis loop of TiO₂/S-InP without PMA treatment is counterclockwise as shown in Figure 10(a), which is from high density of oxide trapped charges [18, 20] in TiO₂/S-InP film without H passivation. The C-V hysteresis loops of PMA-TiO₂/S-InP film are clockwise at 300 and 350°C as shown in Figures 10(b) and 10(c). The mobile ions are responsible for the C-V clockwise hysteresis loop due to the decrease of oxide trapped charges from film quality improvement. The C-V hysteresis loop changes back to counterclockwise at 400°C as shown in Figure 10(d). It is dominated by oxide trapped charges due to the break of H bonds and hence the loss of the passivation function at higher PMA temperature [30, 31]. The sum (N_{ot}) of oxide trapped density and mobile ion density can be derived from the difference of flat-band voltage (ΔV_{FB}) of the C-V hysteresis loops measured at high frequency [20]. The formula is as follows:

$$N_{\rm ot} = \frac{-\Delta V_{\rm FB} C_{\rm ox}}{Aq},\tag{1}$$

where C_{ox} is the oxide capacitance, A is the contact area (7.07 × 10⁻⁴ cm⁻²), and q is magnitude of an electron charge. In *C*-*V* measurement, the bias scans first from accumulation region to inversion region (forward scan) and then back to accumulation region (backward scan). ΔV_{FB} is defined as the difference of V_{FB} between the forward scan and the backward scan. So, the polarity of oxide trapped charge is negative and that of mobile ion charge is positive. The N_{ot} of PMA-TiO₂/S-InP as a function of PMA temperature is shown in Figure 11. The lowest N_{ot} is 7.18 × 10¹⁰ C/cm² at the PMA temperature of 350°C.

The C-V characteristics of TiO₂/GaAs, TiO₂/S-GaAs and PMA-TiO₂/S-GaAs are shown in Figure 12. The C-V characteristics of TiO₂/GaAs show a stretch-out phenomenon under negative bias as shown in Figure 12(a). It is from the high D_{it} due to the existence of native oxides on GaAs surface. The breakdown at higher negative bias is from the higher leakage current resulted in the grain boundary of TiO₂ polycrystalline structure. Figure 12(b) shows the C-V characteristics of TiO₂/S-GaAs. The capacitance in the accumulation region is high due to the improved interface quality and the capacitance decay at higher negative bias is due to the high leakage current. Sharp C-V curves PMA-TiO₂/S-GaAs can be obtained after PMA treatments at 300, 350, and 400°C as shown in Figures 12(c), 12(d), and 12(e), respectively. The ideal C-V curve is also shown in the figure as a reference. The work function difference ($\Phi_{ms} = -1.31 \text{ V}$) of metal (Al) and semiconductor (GaAs) is taken into account. The optimized PMA temperature is 350°C as shown in Figure 12(d), in which the stretch-out phenomenon and the flat-band voltage shift are minimized. The dielectric constants and effective oxide charges of PMA-TiO₂/S-GaAs

FIGURE 13: (a) Dielectric constant and (b) effective oxide charges of TiO₂/S-GaAs and PMA-TiO₂/S-GaAs at different PMA temperatures.

FIGURE 14: C-V hysteresis loops of (a) TiO₂/S-GaAs, (b) PMA at 300°C, (c) PMA at 350°C, and (d) PMA at 400°C.

films as functions of PMA temperature are shown in Figures 13(a) and 13(b), respectively. The dielectric constant and the effective oxide charges can reach 66 and 1.86 $\times 10^{12}$ C/cm² at the PMA temperature of 350°C.

Moreover, the *C*-*V* hysteresis loops as a function of PMA temperature are shown in Figure 14. The *C*-*V* hysteresis loop of TiO₂/S-GaAs without PMA treatment is counterclockwise as shown in Figure 14(a). The *C*-*V* hysteresis loops of PMA-TiO₂/S-GaAs film are clockwise at 300 and 350°C as shown in Figures 14(b) and 14(c). The *C*-*V* hysteresis loop changes back to counterclockwise at 400°C as shown in Figure 14(d). These *C*-*V* behaviors are similar to InP. The N_{ot} of PMA-TiO₂/S-GaAs as a function of PMA temperature is shown in

Figure 15. The lowest $N_{\rm ot}$ is 5.7 $\times 10^{10}$ C/cm² at the PMA temperature of 350°C.

The D_{it} of TiO₂/S-InP and PMA-TiO₂/S-InP at different PMA temperatures is shown in Figure 16. The lowest D_{it} is 7.13×10^{11} cm⁻² eV⁻¹ at the energy of 0.67 eV from the edge of valence band. The D_{it} of TiO₂/S-GaAs and PMA-TiO₂/S-GaAs at different PMA temperatures is shown in Figure 17. The lowest D_{it} is 5.96×10^{11} cm⁻² eV⁻¹ at the energy of 0.7 eV from the edge of valence band. The PMA temperature of two samples was fixed at 350°C.

Table 1 shows the comparisons of electrical characteristics by $PMA(350^{\circ}C)$ -TiO₂/InP and $PMA(350^{\circ}C)$ -TiO₂/ GaAs. Form this table we can clearly recognize that electrical

FIGURE 15: Oxide trap density and mobile ion density of TiO₂/S-GaAs MOS structures as a function of PMA temperature.

FIGURE 16: Interface state densities of TiO₂/S-InP and PMA-TiO₂/S-InP at different PMA temperatures.

characteristics of GaAs are superior to InP. It would be from the fact that the *P* outgas is more serious than that of As due to the higher vapor pressure. It slightly degrades the interface and film quality of InP MOS structure. PMA and $(NH_4)_2S$ treatments highly improve the electrical properties of III-V MOS structures, and dielectric films prepared by lower growth temperature can give more benefits, such as liquid phase deposition and ALD.

4. Conclusions

 TiO_2 films grown on III-V semiconductors with $(NH_4)_2S$ treatments were investigated. With $(NH_4)_2S$ treatment, the interface quality of $TiO_2/III-V$ is much improved. PMA treatment further improves the electrical characteristics. The electrical characteristics of GaAs MOS is better than that of

FIGURE 17: Interface state densities of TiO_2/S -GaAs and PMA- TiO_2/S -GaAs at different PMA temperatures.

InP, which is from the innate character of higher P vapor pressure. There is an interface layer from the inter-diffusion between TiO₂ and substrate. Dielectric films prepared by lower growth temperature for III-V MOS structures can give more benefits.

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