Clinical aspects of trace elements: Zinc in human nutrition – Zinc metabolism

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MM PLUHATOR, ABR THOMSON, RN FEDORAK. Clinical aspects of trace elements: Zinc in human nutrition - Zinc metabolism. Can J Gastroenterol 1995;9(6):327-332. Although zinc has been the most intensely studied trace element, much remains to be learned about its metabolism. Little is known about the normal mechanisms of absorption and transport across the intestinal tract. In addition, numerous unknowns surround the intricacies of bodily zinc homeostasis. Part two of this five-part review presents current views on the normal intestinal absorption, intracellular and extracellular metabolism, transport, excretion and homeostasis of zinc in the human body. The alterations in zinc metabolism that occur with age and changing physiological conditions are also discussed.

Key Words: Absorption, Intestinal transport, Metabolism, Trace element, Zinc

Aspects cliniques et métaboliques des éléments traces : le zinc dans la nutrition humaine

RÉSUMÉ: Bien que le zinc soit l'élément trace le plus étudié, beaucoup reste à découvrir au sujet de son métabolisme. On en sait relativement peu sur son mode d'absorption et son transport normaux dans les voies digestives. De plus, de nombreuses inconnues entourent les relations complexes de l'homéostasie du zinc dans l'organisme. Dans la deuxième partie de cette série de cinq articles, nous passerons en revue l'opinion actuelle sur l'absorption intestinale, le métabolisme intra- et extra-cellulaire, le transport, l'excrétion et l'homéostasie du zinc dans l'organisme humain. Les altérations qui affectent le métabolisme du zinc en fonction de l'âge et de l'état de santé physiologique sont également abordées.

HIS REVIEW IS THE SECOND OF A I five-part series that examines zinc in terms of its biochemistry and physiology, metabolism, dietary require-

states of excess and deficiency.

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ments, nutritional assessment and

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mains unknown about its metabolism. Little is known about its absorption and transport across the intestinal tract. Additionally, the control mechanisms that allow the body to regulate its zinc content and absorption are poorly understood. This review presents an overview of what is currently known about the normal intestinal absorption, intracellular and extracellular metabolism, transport, excretion and homeostasis of zinc in the human body. The alterations in zinc metabolism that occur with age and changing physiological conditions are also highlighted.

INTESTINAL ABSORPTION **OF ZINC**

Absorption of zinc is thought to occur throughout the small intestine (1). Although the highest rate of zinc absorption appears to occur in the jejunum, the duodenum is the first segment exposed to zinc postprandially, and it may thus serve as the primary site for quantitative zinc absorption (1).

Investigators presume that intestinal absorption of zinc is aided by ligands. In the intestinal lumen zinc is likely matched to a ligand with an affinity for this element. The digestion of food leads to the introduction of amino acids, pancreatic secretions, organic acids, phosphates and other cations capable of associating with zinc into the intestinal lumen (2-4). The resultant complexes are then presented to the surface of the mucosal cell.

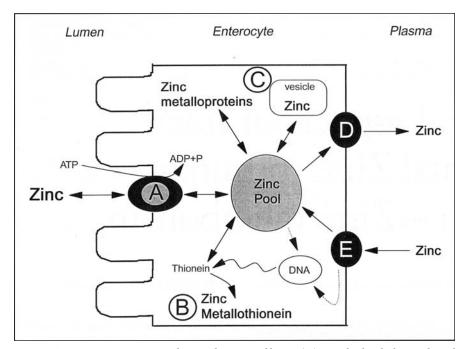


Figure 1) Dietary zinc is transported across the intestinal lumen (A). Newly absorbed zinc is bound by metallothionein and stored (B), used in cell metabolism (C) or released to the plasma (D). When dietary zinc is elevated, plasma zinc and thionein synthesis increase (dashed lines). A large quantity of zinc entering the intestinal cell from the plasma becomes bound to metallothionein (E). Adapted from reference 7

As the intraluminal zinc complexes approach the enterocytes lining the intestinal mucosa, they contact the glycocalyx covering of the brush border membrane. Few studies have assessed the role of the glycocalyx in zinc absorption. However, two potential scenarios have been proposed: zinc is released from the luminal complexes because of a greater affinity for some surface groups of the glycocalyx; or zinc complexes attach to the glycocalyx and are transported across the mucous layer intact (5).

Events at the surface of the brush border membrane have not been fully defined. Using triple-lumen steady state perfusion technology, researchers have found that, over the range of 0.1 to 1.8 mM of zinc, the rate of zinc absorbed is proportional to the amount infused (1). The rate of zinc absorption plateaus with higher concentrations, suggesting a saturable transport mechanism. Additionally, that study investigated the relationship between the intestinal absorption of zinc and that of glucose and sodium. A 'symport' phenomenon was found for zinc and glucose absorption (ie, zinc and glucose interact with a common carrier), while an 'antiport' effect was exerted on sodium absorption (ie, zinc decreases intestinal absorption of sodium, and at high concentrations may even cause a net excretion of sodium). Human intestinal zinc absorption is, therefore, likely to be a carrier-mediated transport process.

Intracellularly, a flux of zinc from mucosa to serosa and from serosa to mucosa appears to be in progress at all times (6). Zinc entering the enterocyte equilibrates with the intracellular zinc pool and can be bound by metallothionein and stored until mature cells are sloughed off, used in the metabolism of the absorbing cell or passed through the serosal border to be picked up by the portal bloodstream (4,7) (Figure 1).

A key role for metallothionein, a sulphur-rich metal-binding protein, in zinc homeostasis was first suggested by Richards and Cousins (8). The authors hypothesized that when metallothionein was synthesized inside the intestinal cell, it could trap zinc and make it unavailable for transfer to the circulating plasma. The amount of metallothionein present in the mucosa was found to be negatively correlated

to zinc absorption, thereby exercising homeostatic control of zinc metabolism. Other reports do not support this concept. Jackson et al (9) refuted the earlier hypothesis when they gave ⁶⁵Zn by gavage to rats and followed the distribution of radioisotope over time. They found that both control and zinc-depleted rats retained the same amount of ⁶⁵Zn in the intestinal wall, even though more ⁶⁵Zn appeared in the carcasses of the zinc-depleted rats. When large amounts of zinc were injected intraperitoneally, zinc absorption decreased after 6 to 24 h, possibly because of the synthesis of metallothionein. Subsequent studies by Cousins (10) have shown that relatively large quantities of zinc are needed to induce metallothionein production, and this event is not likely to happen under physiological levels.

SYSTEMIC TRANSPORT OF ZINC

Approximately 50% of zinc in the plasma is freely exchangeable and loosely bound to albumin; 7% is bound to amino acids such as histidine and cysteine, and the remainder is bound to macroglobulins and other serum proteins (2,11). Alpha2-macroglobulin comprises a tightly bound pool of plasma zinc. The metabolic fate and the physiological significance of zinc circulating with this protein are not clear. Because it is an acute-phase protein that is induced during inflammaalpha2-macroglobulin's zinc-binding capacity may be related to host immune functions (12). A small portion of zinc, about 0.1 to 0.2% of total plasma zinc, is ultrafilterable (12). Ultrafilterable zinc could act as a donor for a high affinity, cellular zinc transport system, or it could represent an intermediate between state protein-bound zinc and that marked for cell uptake (12).

Plasma is the major route of zinc transportation within the body. The plasma's zinc content is approximately 1 μ g/mL, and in most tissues ranges from 11 to 150 μ g/g (13). It is, therefore, obvious that an active or facilitated uptake of zinc by all tissues must occur, but little is known about this

process. Even less is known about the transportation of zinc out of the tissues into the plasma. Certain tissues (eg, gastrointestinal mucosal cells) secrete zinc as part of their normal physiological function. It also appears that a turnover of zinc in other tissues results in a loss of zinc from the cells via unknown mechanisms. It is certain, however, that a constant renewal/exchange system is required to maintain sufficient levels of cellular zinc (10).

Under conditions of catabolism, loss of zinc from the tissues may be excessive. For example, during periods of muscle catabolism, a large loss of zinc from skeletal muscle results in increased urinary and fecal excretion of zinc (13). This elevated urinary zinc may be related to the loss of peptides and amino acids from the muscle during breakdown, which could increase the ultrafilterable 'pool' of zinc in plasma. Another possible explanation is that zinc could be associated with myoglobin and thus be lost as a result of the myoglobinuria that accompanies severe muscle breakdown (13).

Approximately 30 to 40% of the zinc that enters the hepatic venous supply is removed by the liver, from which it is subsequently released back into the blood (14). Circulating zinc is incorporated at differing rates into various extrahepatic tissues, all of which have different rates of zinc turnover. Uptake of zinc by the central nervous system and the bones is relatively slow, and this zinc remains firmly bound for long periods. Zinc is incorporated in the hair and thus becomes unavailable to the tissues. The most rapid accumulation and turnover of zinc occurs in the pancreas, liver, kidney and spleen (14). Uptake and exchange of zinc by red cells and muscle is much slower.

The kinetics of zinc absorption and elimination follow a two-component model. The initial rapid phase has a half-life in humans of 12.5 days, and the slower pool turns over with a half-life of approximately 300 days (14). The latter pool appears to include most of the skeletal muscle zinc. Zinc entering the plasma is cleared within hours, largely through entry into the blood

TABLE 1 Zinc balance

	Amount	Percentage of intake	Reference
Intake	10 to 15 mg/day		
Absorption	3 to 5 mg/day	30	2
Excretion			
Feces	9.0 to 14.0+ mg/day*	90	14
Urine	0.3 to 0.5 mg/day	2 to 10	14
Sweat	0.4 to 2.8 mg/day	Variable	22
Semen	0.6 mg/ejaculum [†]	Variable	22
Menses	0.005 to 0.015 mg/day [‡]	Negligible	29

*Upper limit for fecal zinc excretion based upon excess dietary plus endogenous excretion; [†]Seminal zinc excretion dependent upon individual level of sexual activity; [‡]Based upon a 30-day cycle

cells and tissues rather than by excretion from the body.

The major organ involved in zinc metabolism is the liver. The liver cytosol of humans and other mammals contains zinc-binding components of differing molecular weight and lability, the amounts and proportions of which vary with the zinc nutritional status and age of the animal or human (14). When the liver zinc content is increased above normal levels (30 ug/g wet weight), the additional zinc is associated mainly with metallothionein. Short term fluctuations metallothionein-bound zinc levels could be directly related to daily variations in zinc intake (15).

ZINC EXCRETION

Given normal dietary circumstances, the feces are the major route of zinc excretion. A healthy human adult whose zinc balance is in equilibrium, and who ingests 10 to 15 mg zinc/day, excretes approximately 90% of this amount in the feces, with 2 to 10% excreted in the urine (16). Fecal excretion includes the excretion of both unabsorbed dietary zinc and endogenous zinc. The level of excretion of endogenous zinc in the feces varies according to the balance between true absorption and metabolic needs. This variation is one of the primary mechanisms for maintaining zinc homeostasis (14). Weigand and Kirchgessner (17) demonstrated that the endogenous fecal zinc excretion of rats on a zinc depletion diet was so low that it was exceeded by urinary zinc losses, which are not affected by dietary zinc intake in the rat. An increase in endogenous

fecal excretion that correlates positively with dietary zinc increases has been demonstrated in the human (18). The actual amount of endogenous zinc lost from the human body in the feces may range from less than 1 mg to several milligrams daily (14) (Table 1).

Endogenous fecal losses of zinc arise when zinc is secreted from the body into the gut and is not subsequently absorbed. Matseshe et al (19) used a marker perfusion technique to measure the intraluminal quantities of zinc in the gastrointestinal tract of healthy adult men who consumed different test meals. Over the 5 h following the meal, between 144 and 300% of the amount of zinc ingested was recoverable through aspiration done at the ligament of Treitz. The source of this additional zinc (2.2 to 4.8 mg) was the meal-stimulated secretion of pancreatic fluid. The researchers concluded that an enteropancreatic circulation existed in humans. The pancreatic output of zinc (which is six times greater than that of bile) is the largest source of endogenous mineral re-excretion into the intestinal lumen (6). Much of this zinc must eventually be resorbed to avoid a negative zinc balance.

The proportion of a dose of ⁶⁵Zn that subsequently appeared in pancreatic, biliary and gastroduodenal secretions in human subjects was not enough to account for the total excreted in the feces (14). It was discovered that the other major pathway of zinc excretion is provided by the intestinal mucosal cells. Zinc within these mucosal cells may include unabsorbed zinc taken up from the intestinal lumen and zinc re-entering the cells from the

plasma under conditions of net efflux from the serosal surface to the lumen. Intestinal metallothionein synthesis, which is sensitive to the body's zinc status, may enhance the value of this route of excretion to the overall process of homeostatic control of zinc levels (14).

The amount of zinc excreted in the urine of healthy adults is about 0.3 to 0.5 mg/day (14) (Table 1). Relatively high urinary zinc excretion rates occur in both term and premature infants, but by two months of age these rates are down to 0.01 to 0.02 mg zinc/kg body weight/day (20). Young children aged 33 to 90 months excrete a mean of 0.187±0.097 mg/24 h (21). In adults, the amount of zinc excreted in the urine varies with the level of intake. In the case of adult men who consumed a severely zinc-deficient experimental diet (0.28 mg/day), urinary zinc was decreased threefold to 0.14 mg/day (22).

Urinary zinc arises largely from the ultrafilterable portion of plasma zinc (23). Net reabsorption of up to 95% of filtered zinc under basal conditions occurs in the distal parts of the renal tubule (23). The amount of zinc excreted correlates with the rate of urine (volume) production and creatinine excretion (24). The modest increases in urinary zinc output seen during pregnancy may thus be due to increased tubular flow rates. In parenterally fed infants, little stool is passed; therefore urinary loss becomes the main route for zinc excretion. Eighty-seven per cent of zinc excretion occurs in the urine of parenterally fed infants compared with 8% in healthy breast-fed infants (20). Adults supported by total parenteral nutrition likewise show increased urinary zinc excretion (25,26). This change may partially be explained by the low fecal output of these patients. Additionally, experimental studies done on various total parenteral nutrition solutions and single amino acid infusions have confirmed that infused amino acids, particularly cysteine and histidine, are potent stimulators of renal zinc excretion (20). Causes of increased zincuria may include an increase in the ultrafilterable portion of

plasma zinc and the prevention of zinc resorption through its being bound to amino acids secreted into the renal tubule (23).

Healthy adults living in a temperate North American climate have been reported to lose from 0.4 to 2.8 mg of zinc daily in sweat (22). Studies have reported that whole body sweat losses are related to dietary intake. When intakes of zinc were 8.3 mg/day, young men lost an average of 0.5 mg zinc in sweat daily; losses were reduced to 0.24 mg/day on a marginal zinc intake and increased to 0.62 mg/day when subjects were repleted with a zinc intake of 34 mg/day (27). This reduction of sweat losses during times of dietary deficiency may represent a mechanism for the conservation of zinc. Normal sweat contains 1.15±0.03 µg zinc/mL, approximately 75% of which is in the cell-free portion, with the remainder associated with desquamated epithelial cells (28).

Menstrual losses of zinc are negligible, with average losses of between 0.1 and 0.5 mg zinc occurring per period (29). This represents a loss of about 0.005 to 0.015 mg/day over a typical 30-day cycle, which is a minute fraction of the normal zinc intake.

Despite the high concentrations of zinc found in the prostate and prostatic secretions, very few experiments have investigated seminal zinc losses in men. Baer and King (22) found that seminal zinc losses were roughly comparable with the amounts lost daily in sweat and urine, and were responsible for 20% of the total endogenous zinc losses experienced by their subjects. One seminal emission increased the loss by 0.6 mg/ejaculum/day. The amount of zinc lost via this route was quite variable within and among individuals. In addition, losses depend on the individual level of sexual activity.

ZINC HOMEOSTASIS

The relatively constant levels of zinc found in the body, even though intake varies greatly over time, suggest that an efficient mechanism for whole body homeostasis exists. Homeostasis is maintained by control of gastrointestinal absorption and bodily excretion of zinc. Higher absorption in cases of

lower body zinc load and increased excretion during times of high zinc intake both contribute to zinc homeostasis. Although changes in urinary zinc output contribute little to the maintenance of whole body homeostasis given normal dietary intakes, during times of overt zinc depletion, a major decrease in the urinary zinc output occurs (30).

An increase in zinc absorption rates is a well-described gastrointestinal response to a diet low in zinc, but decreases in the rate of gastrointestinal zinc excretion are also known to occur in response to such a diet (22). In contrast, an increase in dietary zinc intake results in both a reduced fractional absorption of zinc and an increase in the gastrointestinal excretion of zinc (31).

When these primary homeostatic mechanisms are not sufficient to handle large dietary reductions or excesses of zinc, a second line of defence enters. A large excess of zinc leads to tremendous losses via nonphysiological routes (eg, hair) (13), while a highly zincdeficient diet causes a selective loss of zinc from certain body fluids and tissues (13). Animal studies have shown that plasma zinc levels can drop rapidly after the introduction of a low zinc diet (32) and that liver zinc levels reflect recent dietary intakes (33). However, tissues that show symptoms of acute zinc deficiency (eg, skin, hair and gut) show little initial change in their total zinc content. The rapid fall in plasma zinc levels that follows the introduction of a diet low in zinc may be part of a homeostatic mechanism developed to support preferentially zinc maintenance in tissues susceptible to deficiency symptoms (13). In much the same way, the preferential removal of surplus zinc through incorporation into the hair is likely a method of preventing damage to other tissues.

The mucosal cell appears to play an important role in zinc homeostasis. Mucosal uptake of zinc occurs via two processes: a nonsaturable process not affected by dietary zinc intake, and a saturable process stimulated by zinc depletion. Results of studies of weanling rats provided with a range of dietary zinc from 5.6 to 141 mg/kg showed that, even though zinc intake rose line-

arly, true absorption, apparent absorption and apparent retention each plateaued at about the 40 ppm level (17). At the lowest dietary intakes, a quantitative (100%) absorption of dietary zinc takes place, and fecal zinc is derived entirely from endogenous losses. However, at higher levels of zinc intake, a decrease in true absorption and an increase in endogenous excretion into the gut combine to provide stable bodily retention. Studies of in situ ligated loops of rat intestine infused with zinc concentrations ranging from 1 to 200 µg/mL have displayed saturation kinetics suggestive of a carrier-mediated function (34). Further, at the lower end of zinc infusion levels (1 to 50 µg/mL), mucosal binding took place at specific sites; beyond this range, binding was not specific. Thus, absorption occurs via a saturable process: above a certain level, the capacity of this process is exceeded. Excretion of zinc is the main regulator of zinc homeostasis at higher dietary zinc intake levels (35).

CHANGES IN ZINC METABOLISM WITH AGE AND PHYSIOLOGICAL CONDITIONS

Recommended intake levels of zinc are highest for infants and for pregnant and lactating women. These higher levels take into account the increased demands for zinc experienced by the growing fetus and infant; milk production also requires an increased zinc supply. However, there is also a possibility that the efficiency of intestinal zinc absorption is increased during these times of rapid growth and production. Studies by Davies and Williams (36) on pregnant rats found that the absorptive capacity of the small intestine increased during periods of pregnancy and lactation. Zinc absorption per duo-REFERENCES

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denal loop was found to increase after the 12th day of pregnancy and to continue to rise during lactation. During the latter part of pregnancy, zinc absorption was found to increase specifically (ie, in contrast to a general increase in nutrient absorption), probably to meet the increased needs of the fetus because maternal zinc retention levels were unchanged. Mucosal hypertrophy was responsible for the general increase in nutrient absorption, including zinc, during lactation. Swanson et al (37), using stable isotopes, found no significant difference in the zinc absorption of pregnant and nonpregnant women. It is possible that a small increase in zinc absorption rates during human pregnancy may be sufficient to store enough zinc to cover the needs of the fetus, while the rat, with its many fetuses and short pregnancy term, may require a much faster zinc accrual rate (5).

Little information is available on the effect of ageing on zinc metabolism in humans. It is, however, well known that the newborn animal absorbs zinc to a higher degree than does the older animal (17). Studies done on rats of different ages suggested that the transport system displayed a higher affinity for zinc at a young age. Total zinc uptake was shown to increase with age, and values were higher for younger versus adult rats. Another study showed that the intestinal absorption of ⁶⁵Zn decreases with age in rats (38). In humans, the retention of zinc with age has been reported (39). The values for zinc balances reported for children and young adults, compared with zinc balances of older persons, indicate that zinc metabolism changes with age (40). Decreases in both plasma and tissue levels of zinc have been reported, mainly in individuals with pathological zinc metabolism. Adv Nutr Res

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conditions. However, these persons are usually older individuals. As ageing and various disease states may both affect zinc status, it becomes important to be able to distinguish between the effects of the two variables. Very little information exists on zinc metabolism in the older adult population. Among the parameters of zinc metabolism that may possibly change with age are intestinal absorption of zinc, endogenous excretion, tissue retention and zinc balance, plasma levels and exchangeable zinc pool status. Should it be possible to distinguish significant changes in zinc metabolism associated with age, corrective measures may aid in reversing these changes; such corrective measures may also prevent the development of disease states in which abnormal zinc statuses are involved.

CONCLUSIONS

Although medical researchers have a more comprehensive understanding of the role of zinc in human nutrition and homeostasis than they do for any other trace element, there is a considerable lack of knowledge about normal zinc metabolism and the metabolic sources of the pathological changes that occur during times of zinc deficiency. The relatively constant control exerted over zinc levels in the tissues and body fluids, maintained even with diverse dietary intakes, suggests that efficient mechanisms work to control whole body homeostasis. Investigation of the factors responsible for these adaptive controls will likely provide long-anticipated insights into the complexities of zinc metabolism, including the ways in which human physiological processes change as the human body ages, and faces the challenges of pregnancy, growth and disease.

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